



पुर्ना International School

Shree Swaminarayan Gurukul, Zundal

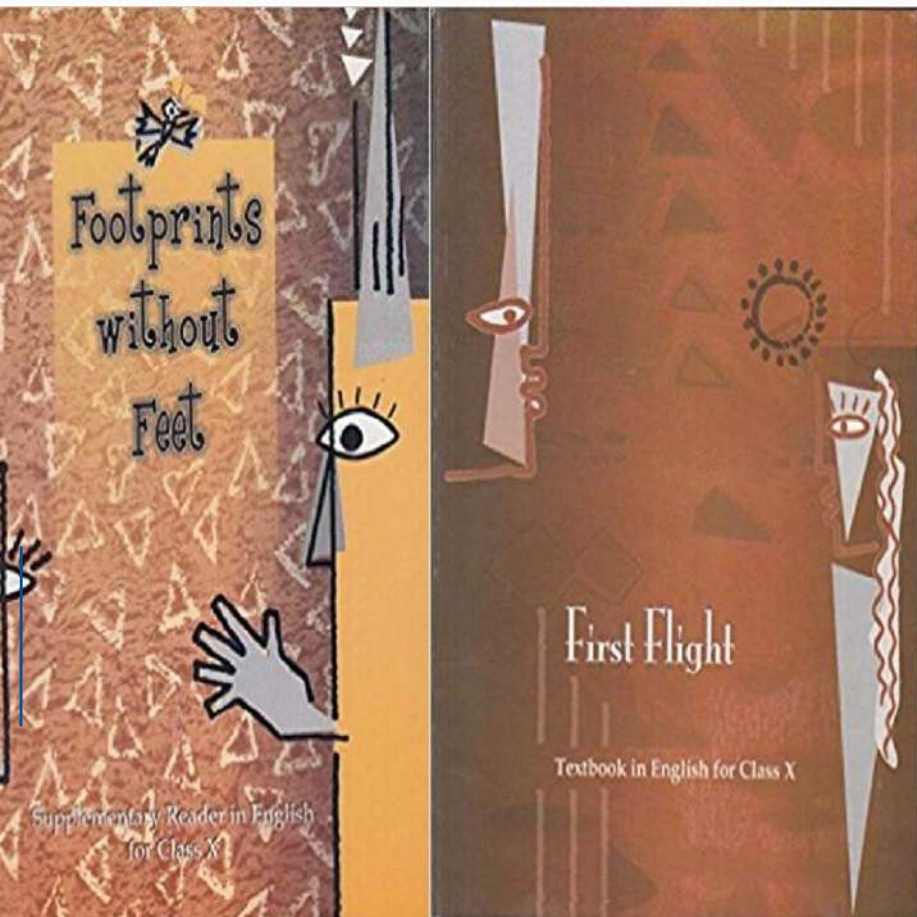
**Knowing the learning process
Grade X Teaching methodology
September and October 2019**



Subjects

•Subjects	Code
•English	184
•Hindi	085
•Mathematics	041
•Social studies	087
•Science	086

Glance at lessons of September and October 2019



Prose

- L6 Glimpses of India
- L7 Mujbil The Otter

Poetry

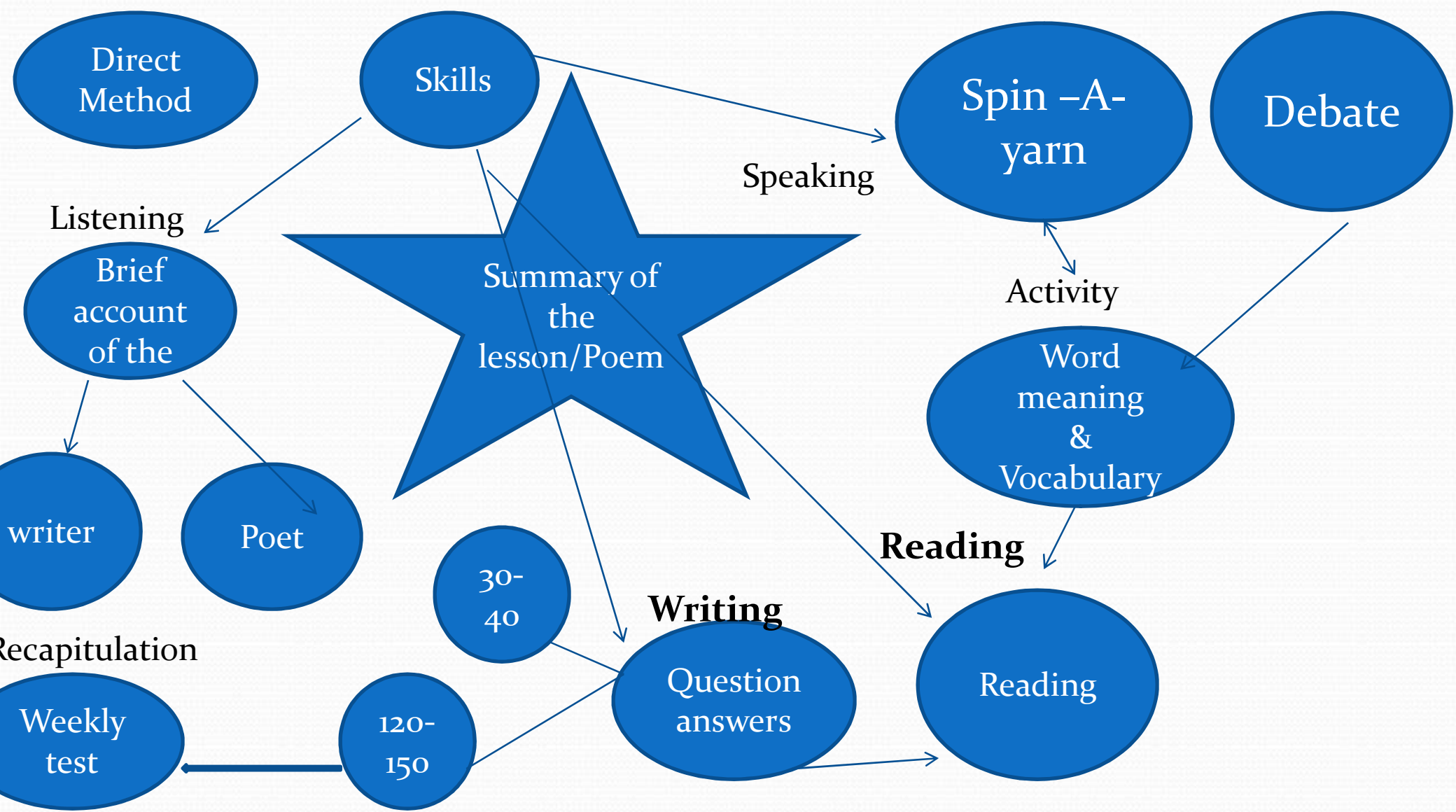
- L 7 Animals
- L 8 The Trees

Supplementary

- L 7 The Necklace
- L8 The Hack Driver



Teaching Methodology



Teaching Aids

Flow chart

Supp. L-6 – The making of scientist

Only child of parents



Fascination for butterflies



Guided by guru Uргу Hart



Started doing experiment



Started winning award



Realised importance of cell



Won the first prize

Became good debater

Sight words
Pronunciation
Sentences

Tredrick
Weiherer
Ebright

Uргу
incredible



ROBERT W. PETERSON

Flash
Cards

tube
eo

<https://www.youtube.com/watch?v=4-vPTbOSokQ>

Teaching Aids

Flow
chart

Prose L-6
'Glimpses of
India'

Baker of Goa-
making of Bread loaves



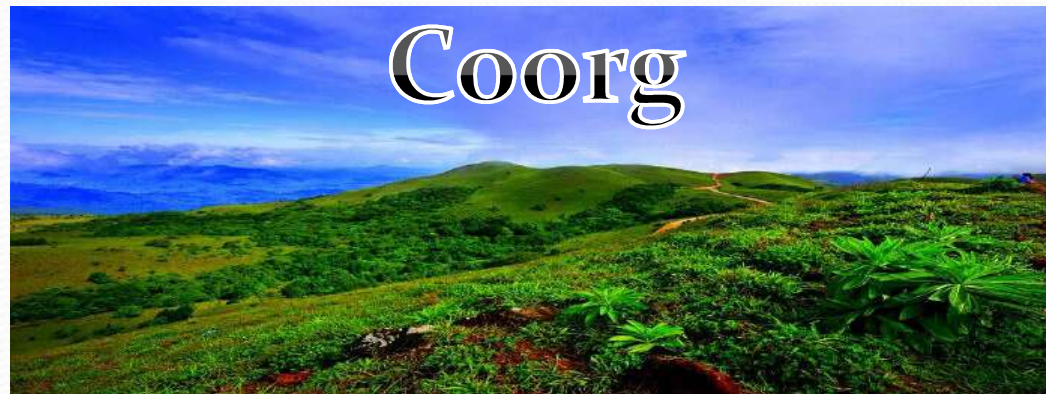
Bakers of Goa

Coorg- heavenly
place, good
hospitality

Tea from Assam,



Tea of Assam



Flash
Cards



Lucio Rodrigues

Sight words
Pronunciation
Sentences

Jingling Kabai
Mysore Brahma
Bylakuppe Pranjol

YouTube
video

https://www.youtube.com/watch?v=u_1Fyczjgm8



PAPER STYLE

A Section	Reading skills	20 Marks
B Section	Writing with grammar	30 Marks
C Section	Literature TB & Extended Reading	30 Marks

Section A Reading **20 Marks**

Q1 : A factual passage 300-350 words 8 Marks

Q2 :A discursive passage 350-400 words with 4 short answer type to test vocabulary. 12 marks

Section B Writing and Grammar **30 marks**

Q3 : Writing an article/ descriptive paragraph(person place event /diary entry) in about 100-150 words 8 Marks

Q4: Writing a short story based on given outline 10 Marks

Q5 : Gap filling with one or two words to test preposition, articles, conjunctions and tenses 4 Marks

Q6 : Editing / Omission 4 Marks

Q7 : Sentence reordering / sentence transformation in context 4 marks

Paper style

Section C

30 Marks

- Q8 : One out of two extract from prose/ poetry/ play . Four very short answer qs **4 Marks**
- Q9: Five short answer type qs. From Beehive and Moments (3 from Beehive 2 from moments) 30-40 words. **10 Marks**
- Q 10: One out of 2 long answer type qs from Beehive to assess creativity , imagination beyond the text book (100 – 150 words) **8 Marks**
- Q11: One out of two long answer qs . From Moments on theme or plot interpretation beyond the text or character sketch **8 Marks**



सपत्नी

भाग १

कविता 10 के लिए लिखी
(शिक्षण सामग्री) जी. ए. ए. ए. ए. ए.





पर्वत प्रदेश में पावस

रुस्किन बॉन्ड का
Class III English (अवधि)

कवि- परिचय

सुमित्रानंदन पंत

NCERT

मूल नाम: गोसाँई दत्त

जन्म: सन् 1900, कौसानी, जिला अल्मोड़ा (उत्तरांचल)

प्रमुख रचनाएँ: वीणा, ग्रंथि, पल्लव, गुंजन, युगवाणी, ग्राम्या, चिंदबरा, उत्तरा, स्वर्ण किरण, कला और बूढ़ा चाँद, लोकायतन आदि

सम्मान: भारतीय ज्ञानपीठ पुरस्कार, साहित्य अकादमी पुरस्कार, सोवियत लैंड नेहरू पुरस्कार, पद्मभूषण

मृत्यु: सन् 1977



ठ-सार

पर्वत प्रदेश में वर्षा ऋतु में प्राकृतिक सौंदर्य कई गुना बढ़ जाता है। वहाँ क्षण-क्षण प्रकृति अपना वेश बदलती-सी नज़र आती है। कभी धूप चमकती नज़र आती है, कभी सूर्य बादलों की ओट में छिप जाता है, कभी प्रकृति का सुहावना रंग दिखाई देता है, तो कभी इतने घने बादल छा जाते हैं कि पर्वत तक अदृश्य हो जाते हैं। मात्र झरनों का शोर सुनाई देता रहता है। अचानक घनघोर वर्षा होने लगती है। निःसंदेह पर्वतों की प्रकृति के ये बदलते दृश्य सुहावने तो लगते हैं। पर्यटकों को आकर्षित भी करते हैं, परंतु पहाड़ों पर रहने वाले लोगों के लिए यह मौसम कठिनाइयों का कारण भी बन जाता है। वर्षा ऋतु में बादलों का फटना, चट्टानों का खिसकना बर्फीले तूफानों का आना एक आम समस्या है, जिसमें गाँव-के-गाँव तबाह हो जाते हैं। फिसलन भरे रास्तों के कारण यातायात व्यवस्था ठप्प पड़ जाती है जिससे रोजमर्रा के लिए आवश्यक सामग्री तक उचित समय पर नहीं पहुँच पाती। चिकित्सा-सुविधाएँ न पहुँचना, संचार व्यवस्था का ठप्प होना, सड़कों का टूटना, ऐसी अनेक समस्याएँ हैं। जिनका सामना इन पर्वतीय अंचल में रहने वाले लोगों को करना पड़ता है।



खन विधि -

काव्यांशों की विस्तृत व्याख्या -

शब्दार्थ

सन्दर्भ

व्याख्या

निष्कर्ष

लघु उत्तरीय प्रश्नोत्तर

नेबंधात्मक प्रश्नोत्तर



परिवर्तन

व्याख्याओं का मौखिक अभ्यास

प्रश्नोत्तर का मौखिक अभ्यास

श्रुतिलेख

साप्ताहिक टेस्ट

Co-ordinate Geometry



What is In this chapter

- Introduction
- Distance formula
- Section formula
- Area of triangle



7 class 10

DISTANCE FORMULA

If $A(x_1, y_1)$ and $B(x_2, y_2)$ are two points then distance AB is found by

SECTION FORMULA

Coordinates of the point $P(x, y)$ which divides the line segment joining the points $A(x_1, y_1)$ and $B(x_2, y_2)$ internally in the ratio $m_1:m_2$ are

AREA OF TRIANGLE

$\Delta ABC = \frac{1}{2} \times \text{base} \times \text{altitude}$

Trigonometry

Made by : Pankaj chakraborty



What is in this chapter

- Introduction
- Trigonometric ratios
- Trigonometric ratios of some specific angles
- Trigonometric ratios of complementary angles
- Trigonometric identities



Chapter 8

In right angle triangle ABC , $\angle B = 90^\circ$,

$\sin A = \text{opposite side} / \text{hypotenuse}$

$\cos A = \text{adjacent side} / \text{hypotenuse}$

$\tan A = \text{opposite side} / \text{adjacent side}$

Trigonometric Identities

$$\sin^2\theta + \cos^2\theta = 1$$

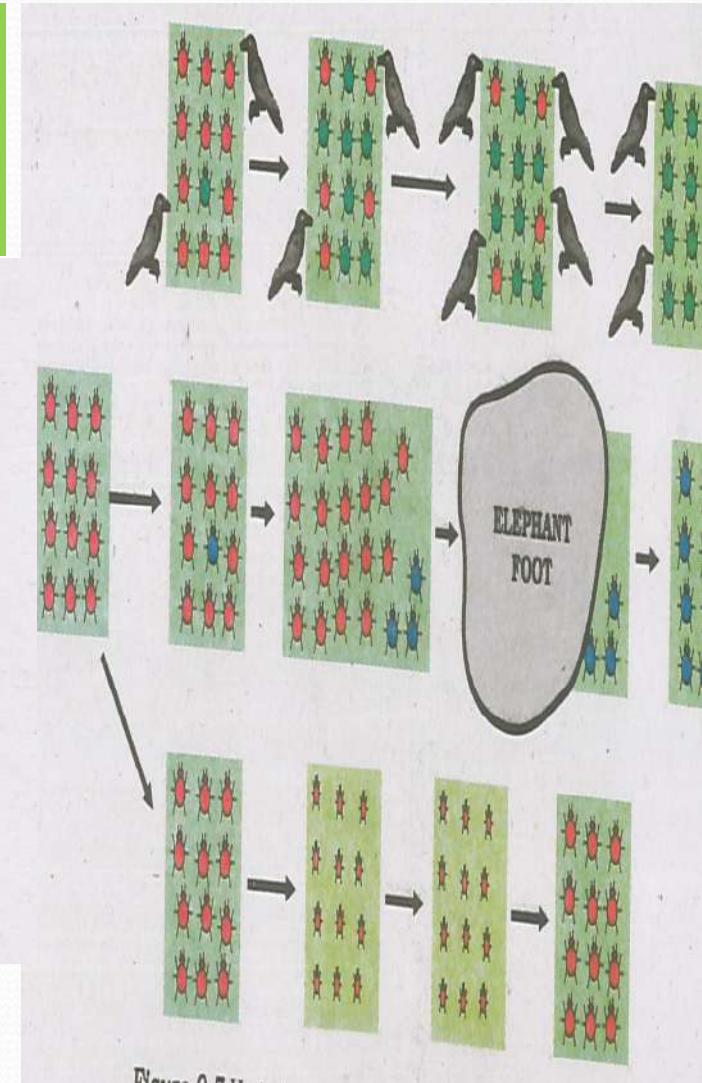
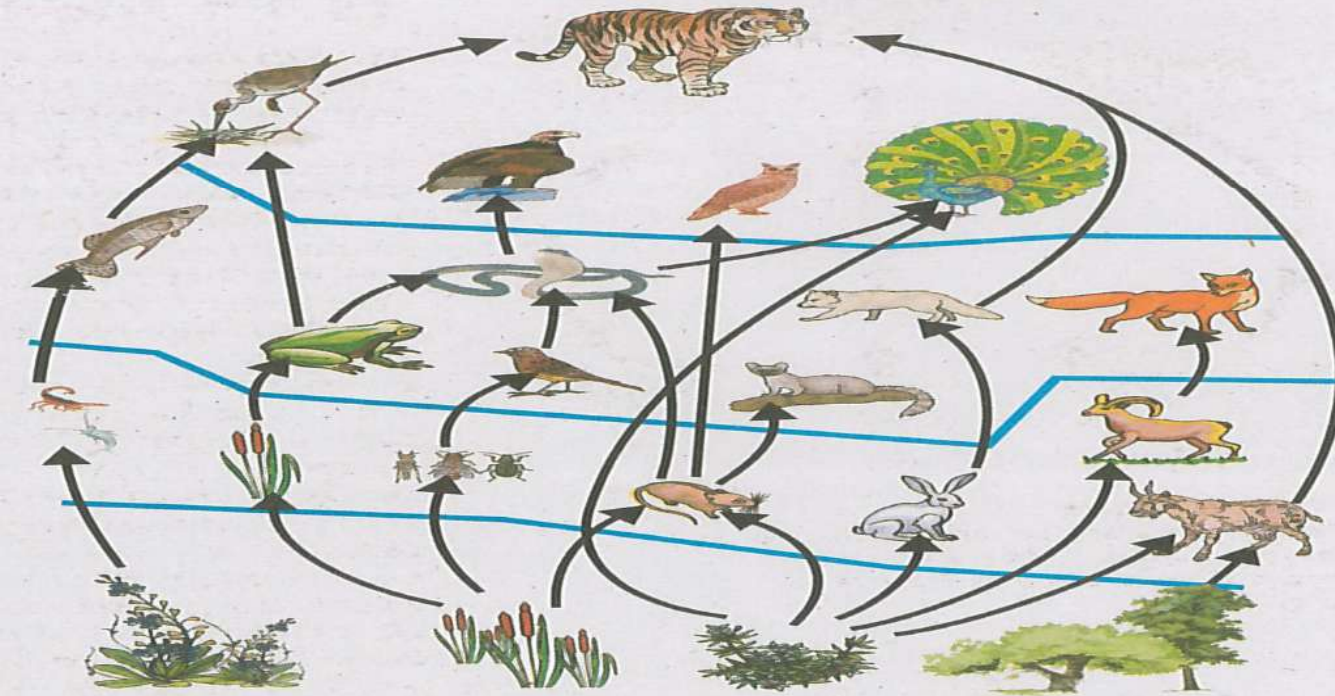
I. $1 + \tan^2\theta = \sec^2\theta$

II. $1 + \cos^2\theta = \operatorname{cosec}^2\theta$

BIOLOGY

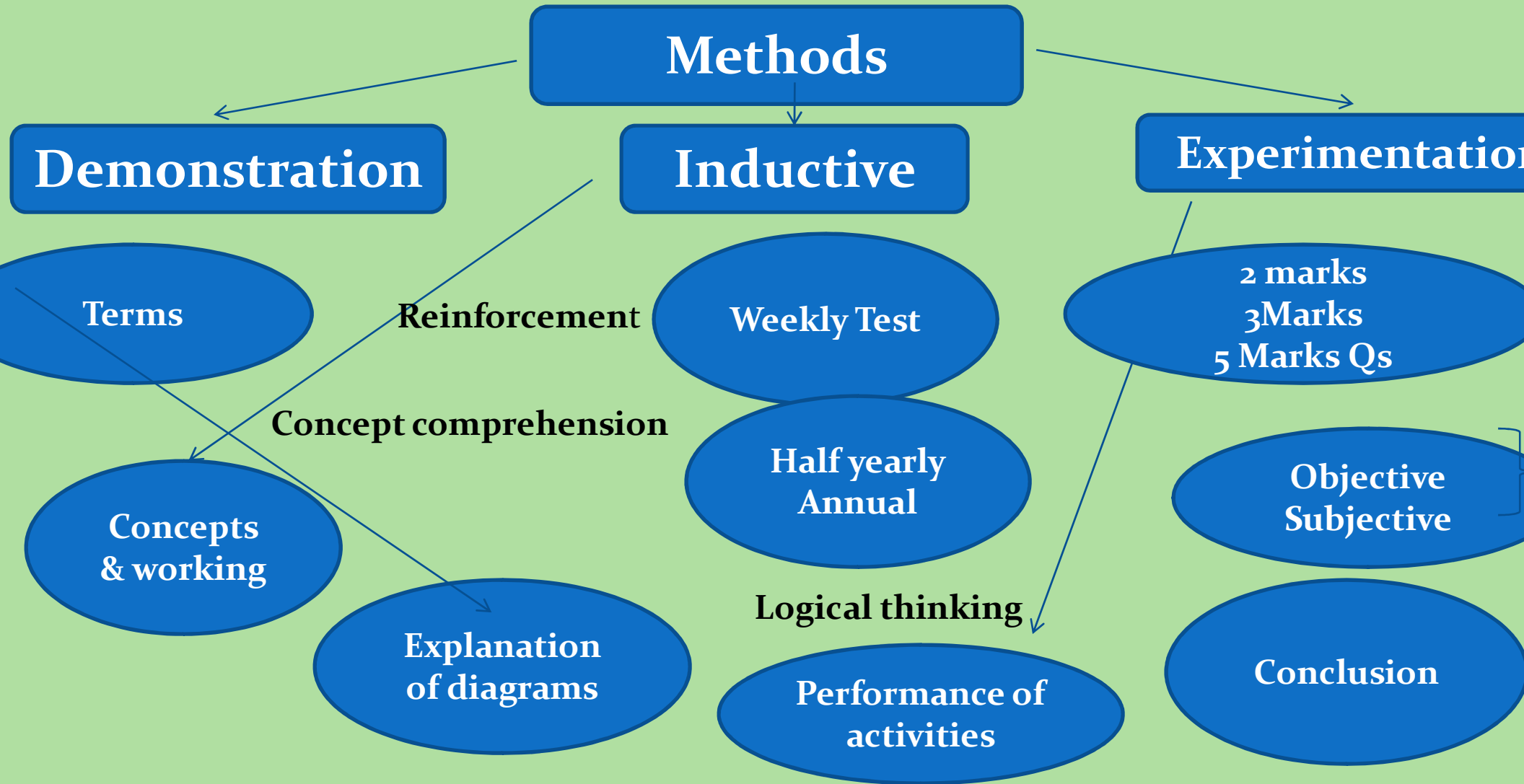
Chapter 9 Heredity and Evolution

Chapter 15 Our Environment



Teaching

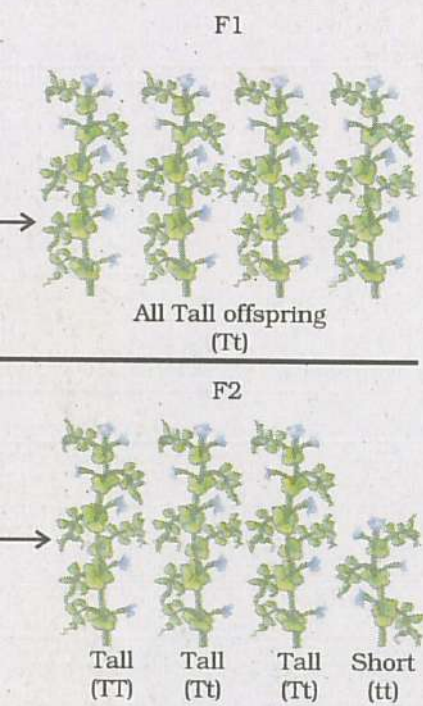
Methodology



Teaching Aids

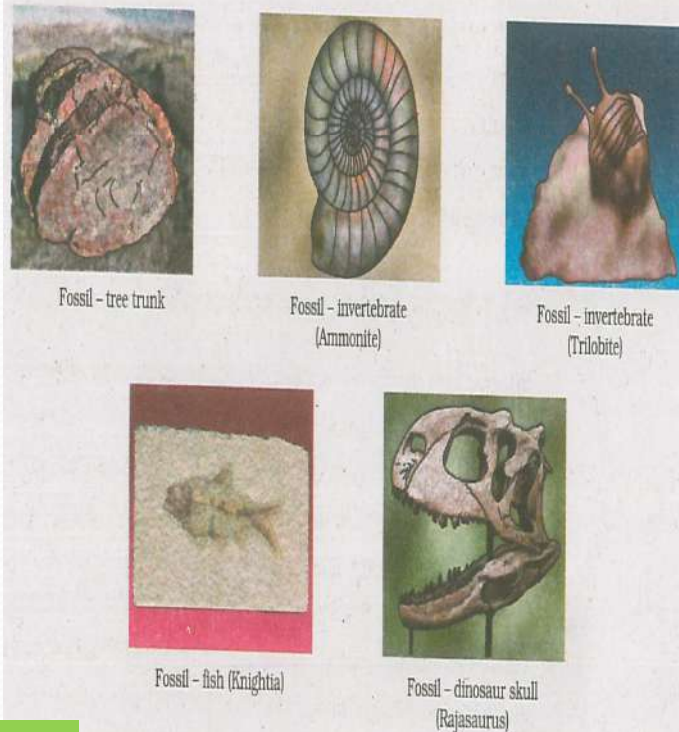
Diagrams

Heredity and Variation



Inheritance of Traits

2) Various Fossils



3) Dinosaur Skull from Narmada

Flash cards

1)

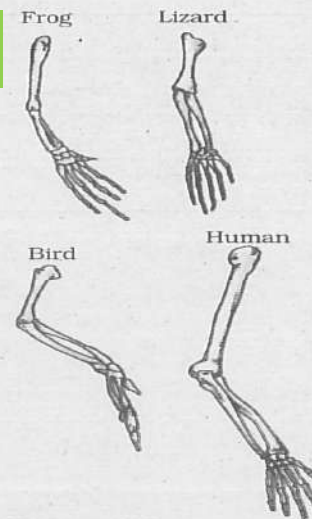
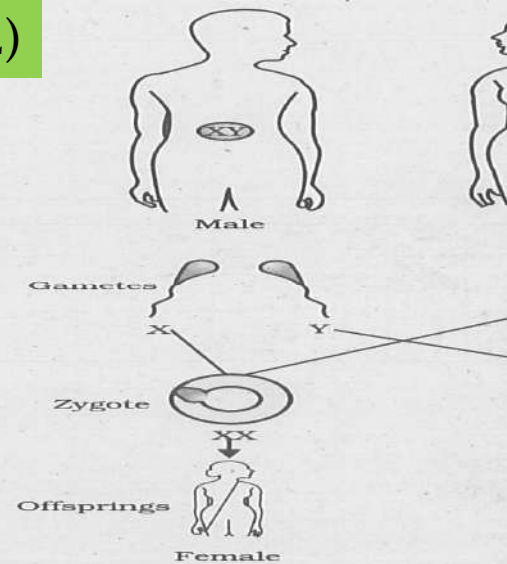


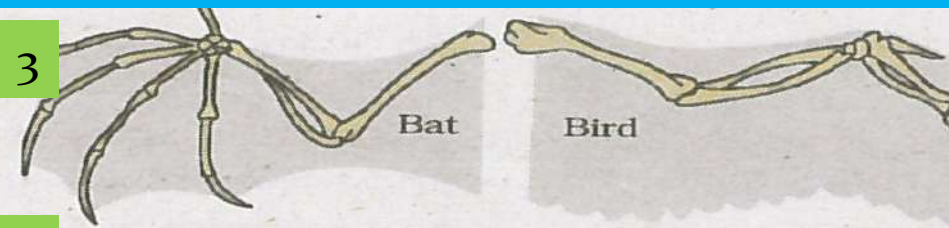
Figure 9.8 Homologous organs

2)



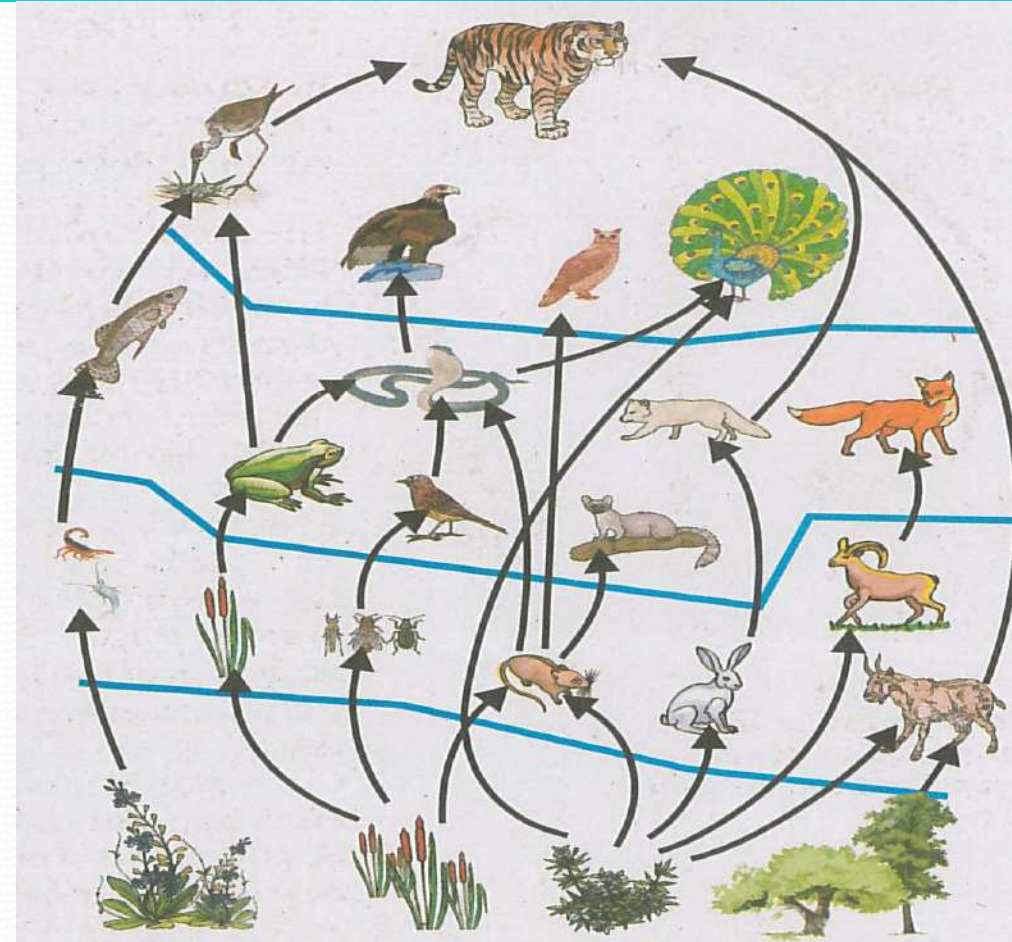
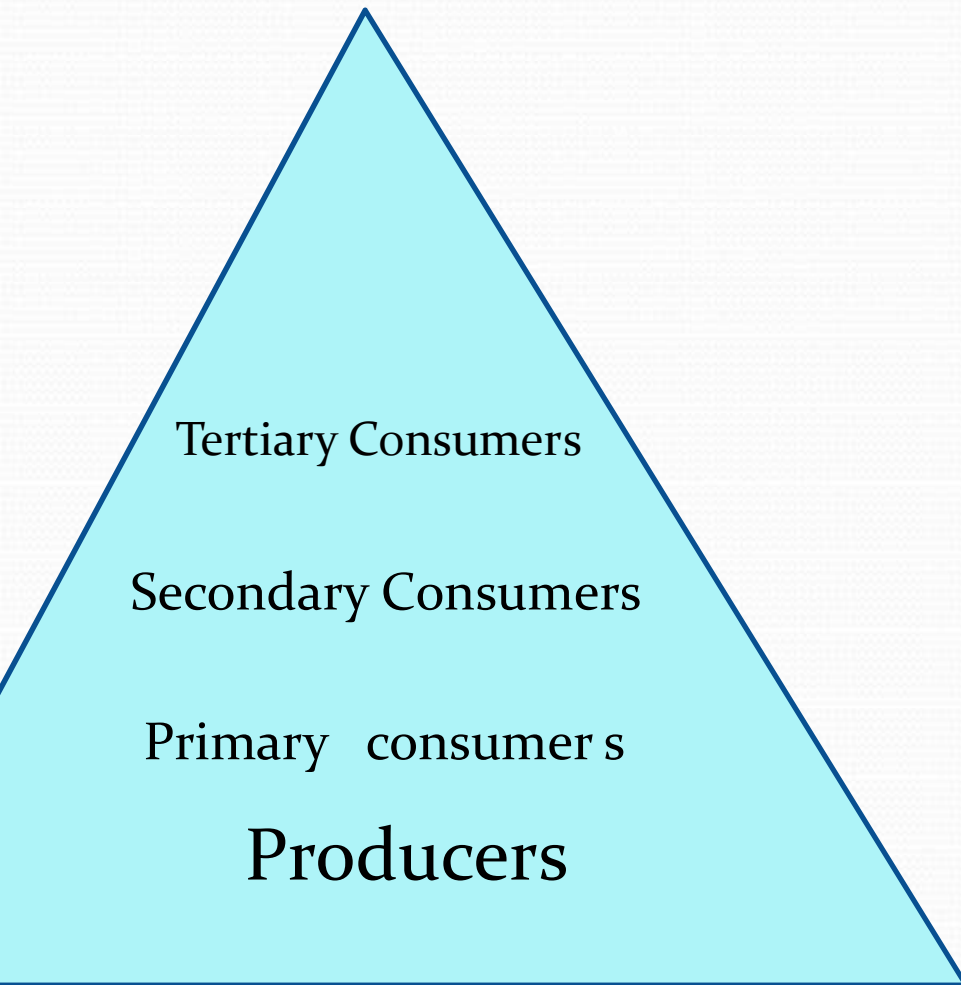
Homologous 2) Sex Determination in human

3)



3) Analogous Organs

L. 15 OUR ENVIRONMENT



Food Web consisting of many chains



REVIEW :September & October COURSES

- How do Mendel's experiment show that traits are inherited ?**
- How does the creation of variation in a species promote survival ?**
- Why are the small number of tigers a cause of worry from the point of view of genetics?**
- What are the problems caused by the non-biodegradable waste that we generate?**
- Why disposable cups made of clay introduced ?**

Teaching material :

1. PICTURES:

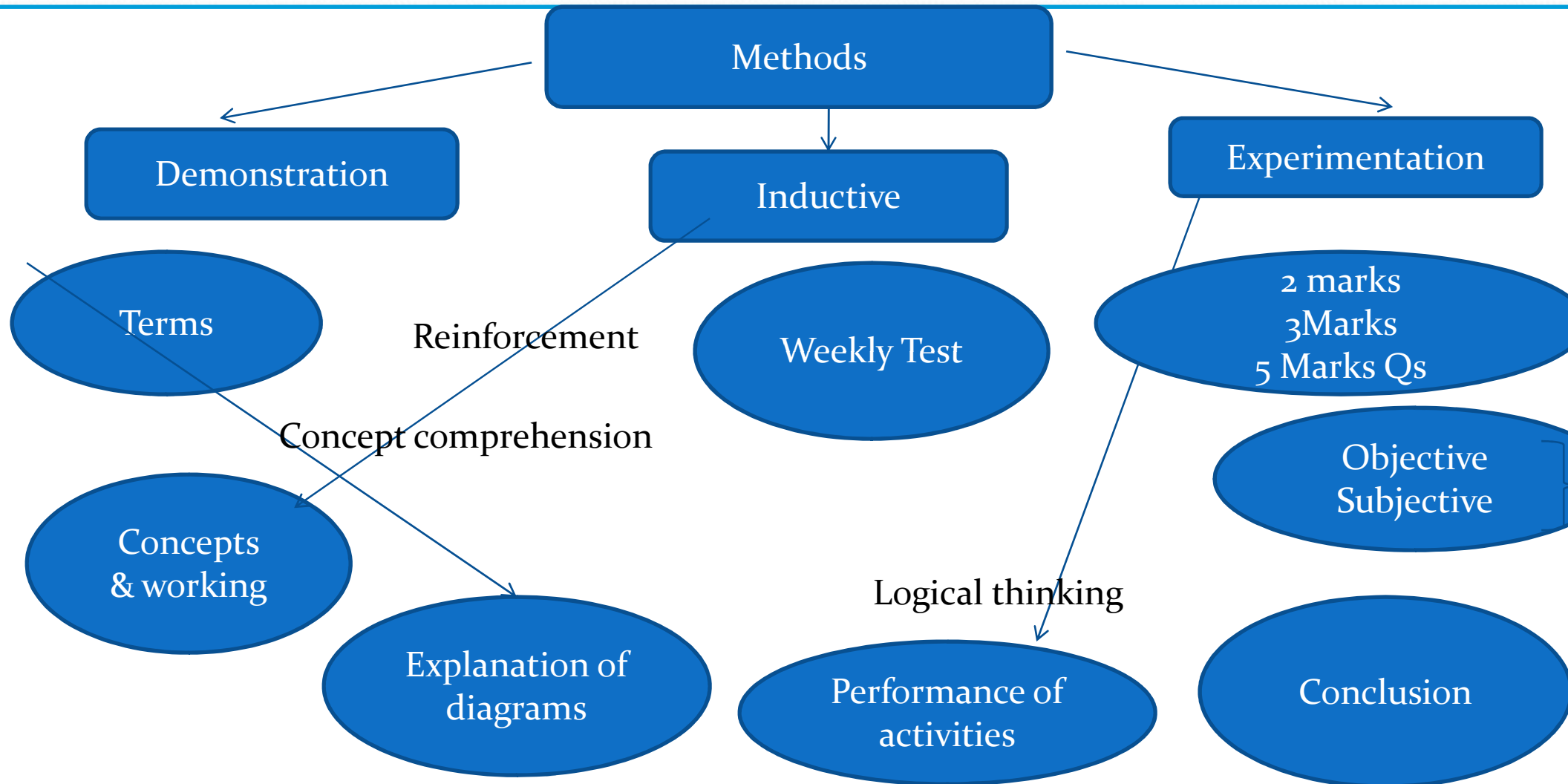
Different forms of money

OLD FORMS OF MONEY



TEACHING

METHODOLOGY



Demonstration and Observation

Demonstration of morphological characters of Bat and earth worm



Continued Teaching aids

Youtube videos

<https://www.youtube.com/watch?v=ZVP3s1KMjUU>

<https://www.youtube.com/watch?v=325VZvdNodk>

<https://www.youtube.com/watch?v=U-g8Y5GvmxU>

<https://www.youtube.com/watch?v=gEk6JLJNgoU>

NCERT Text book

Dr. PS Verma and VK Agarwal

Science Paper style

The question paper comprises of five sections

A,B,C,D and E

All question are compulsory

Internal choice is given in Sections B, C,D and E

Sec.A Q no 1 and 2 in this section are 1 mark each.

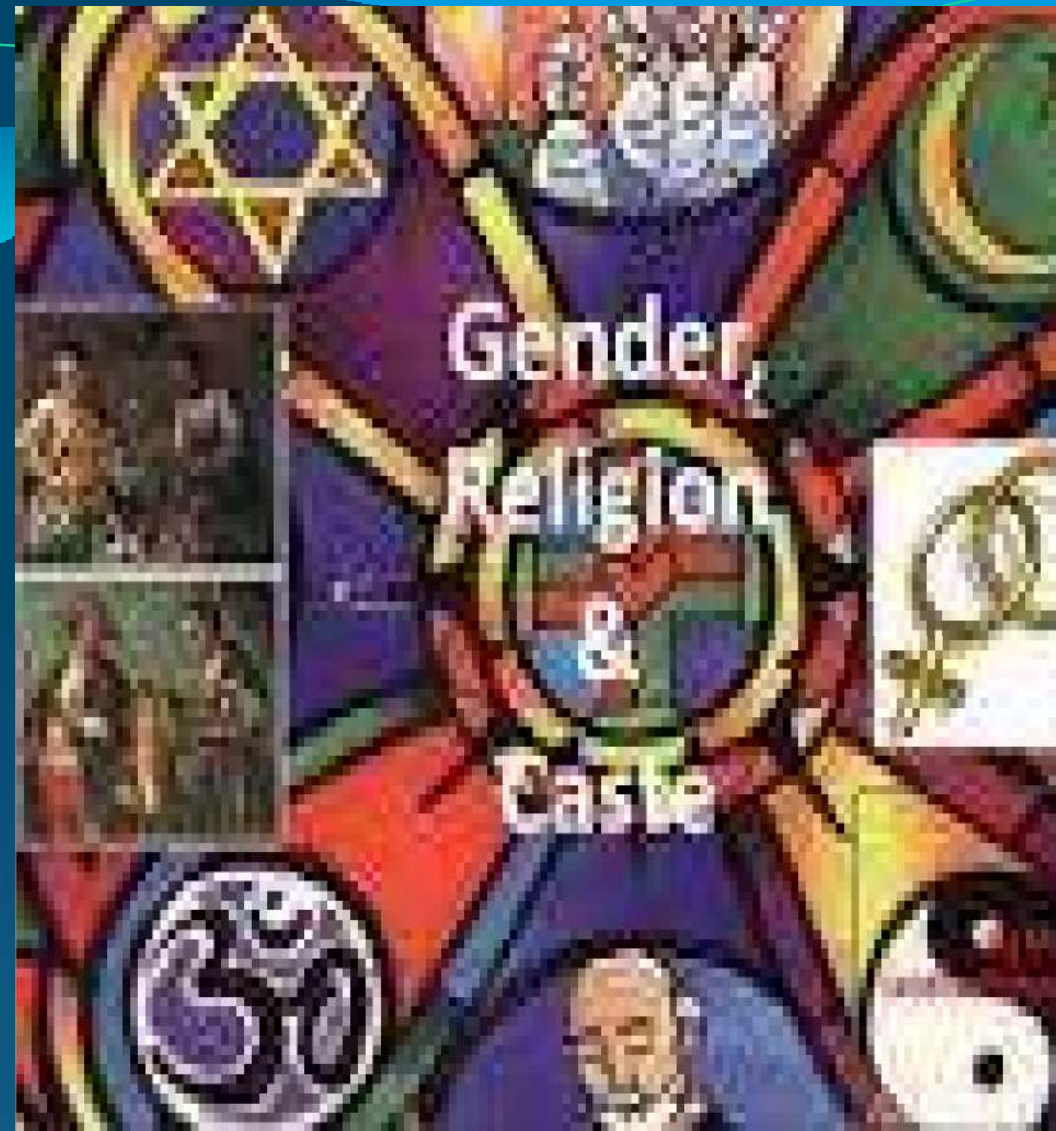
Q. No 3-5 in Section B are 2 marks each to be answered in 30 words

Q. No 6-15 in section C are 3 marks each. They are to be answered in 50 words.

Q. No 16- 21 in Section D are 5 marks each .they are to be answered in 70 words

Q. No 22-27 in Section E are based on practical skills . Each questions is of 2 marks each

Social Science GENDER RELIGION & CASTE



THIS
IS
BEING
A
GIRL
A

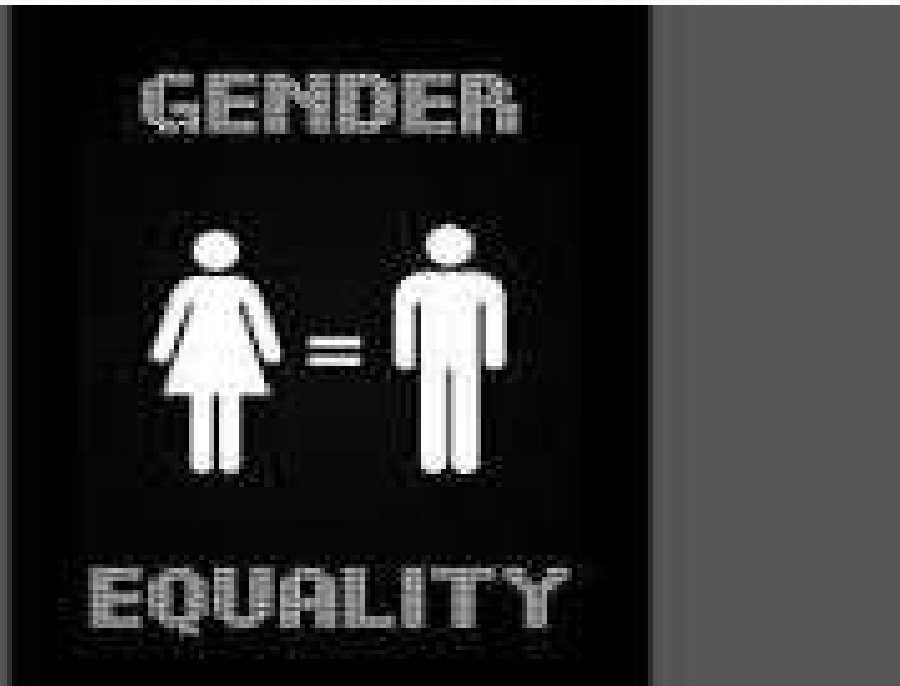


CRIME ✓

DIFFERENT ASPECTS OF LIFE WHERE WOMEN DISCRIMINATED

literacy rate among the women is 54% as compared to 70% among men.

women are paid less than men, even when both do the same work. e.g. in factories, films women are paid less as compared to men.



India, parents prefer to have sons than girls. There are many ways to have the girl child aborted, be

his decline in the sex ratio. The number of girls per thousand boys in our country is 933.



There are reports of various kinds of harassments, intimidation and violence both in rural and urban areas. Women are not safe at home even due to harassment and domestic violence.



FEMINIST MOVEMENT

feminist movements are the movements organized by the women organizations for equality of women in personal and family life.



GLOBALISATION & THE INDIAN ECONOMY

The image features a central globe of the Earth, showing continents in green and oceans in blue. The globe is being held up by several hands of various skin tones, including light, medium, and dark brown. The hands are positioned around the globe, with some fingers pointing towards it, suggesting a collective effort or support. The background is a deep blue, possibly representing the sky or the ocean. The overall composition is centered and balanced, with the text overlaid on the upper portion of the globe.

WHAT IS GLOBALISATION

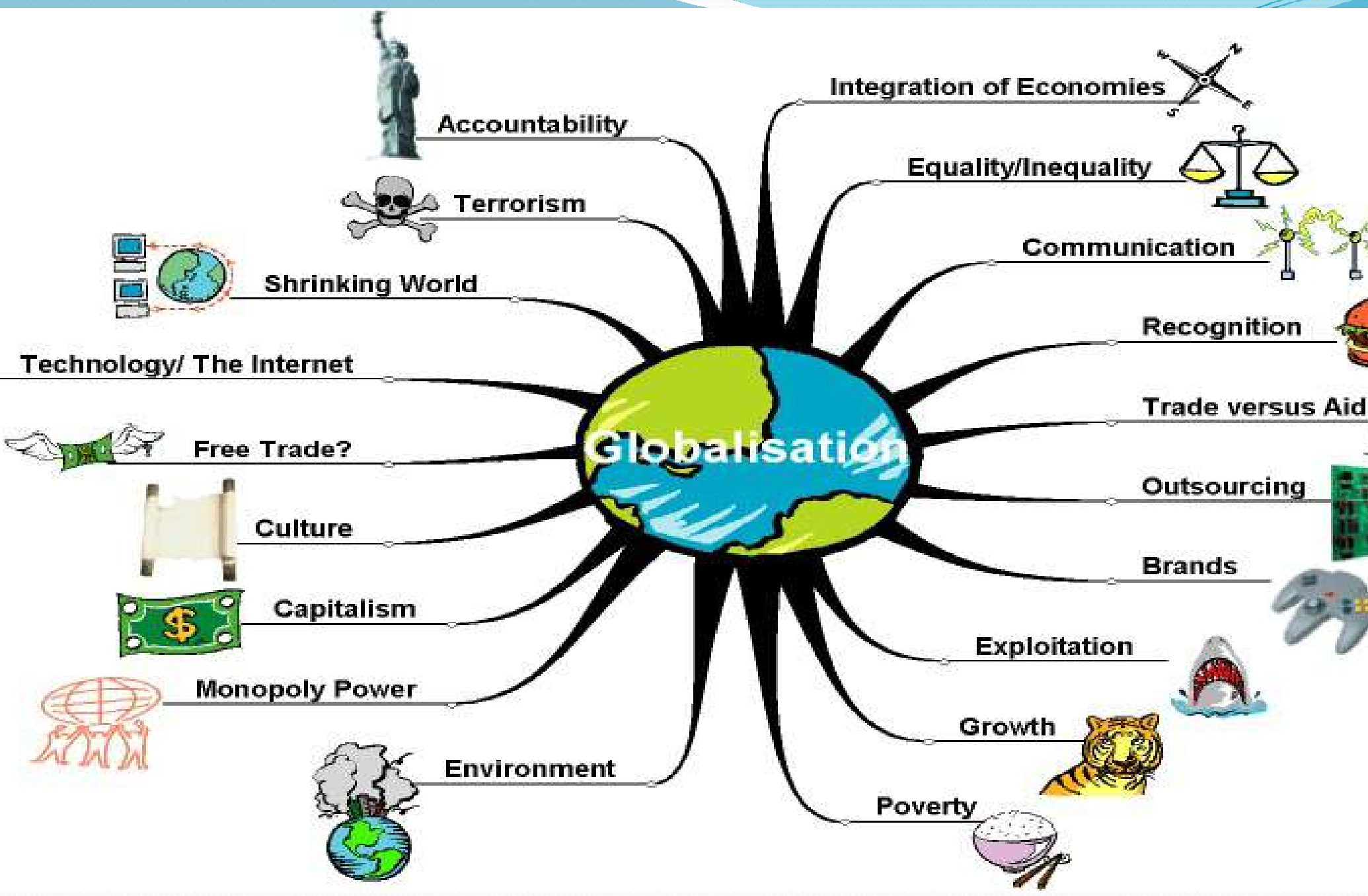
It is the process of rapid integration or interconnection between countries.

- There is one more way through which countries are coming closer and that

Movement of people between countries. People usually move from one country to another in search of jobs or better education. This is also a result of

globalisation.

Globalisation





Impact of Globalisation in India

Greater competition among producers - both local and foreign producers has been of advantage to consumers.

There is greater choice before these consumers who now enjoy improved quality and lower prices for several products.

Foreign investment has increased.



FACTORS THAT HAVE

ENABLED

GLOBALISATION

TECHNOLOGY: Rapid improvement in technology has been the major factor that has stimulated globalisation process. Due to technology there has been improvements in various fields as in :

TRANSPORTATION TECHNOLOGY.

In past fifty years this technological improvements has led to faster delivery of goods across long distances at lower cost.


Containers for transport of goods: have led to huge reduction in port handling costs, increased the speed with which goods can reach markets.



Airlines: the cost of air transport has fallen, this has enabled much greater volumes of goods being transported by airlines.

INFORMATION AND COMMUNICATION TECHNOLOGY:

has played a major role in spreading out production services across countries.



markable improvements have in the areas of
communications, computers & internet.

Telecommunications: facilitated by the satellite
communication devices, facilities as telegraph, telephone
including mobiles, fax are used to
contact around the world, to access the
information instantly, & to communicate in the remote area



Computer and internet: computers have entered almost all the fields.

Internet allows one to share information on almost everything, we can send instant e-mail and talk through video-mail across the world at almost negligible cost.





THESE RAPID TRANSFORMATIONS?

Middle of twentieth century:

Production was largely organized within the countries

What crossed the boundaries was mainly the raw materials, food stuff and finished products.

Trade was the main channel connecting distant countries.

TRADE HISTORY:

Various trade routes connecting India and South Asia to markets in the East and West & extensive trade that took place along these routes.

★ It was trading interest which attracted various trading companies such as East India Company to India.



Foreign Trade & Integration of Markets

Exchange of goods (purchase and selling)
across different countries

Producers in the two countries closely compete
with each other though
they are separated by thousand miles distance

Foreign trade results in connecting
the markets or integration of markets in different
countries.

Goods move from one market to another

Choice of goods in the market rises.

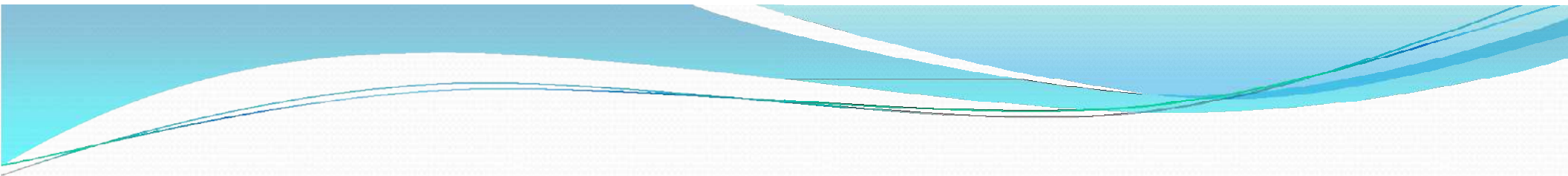
Price disparity reduces i.e. prices of similar goods
in the two markets tend to become equal.



Function or purpose of foreign trade?

Foreign trade creates an opportunity for the producers to reach beyond the domestic markets i.e., markets of their own countries.

Producers can sell their produce not only in markets located within the country but can also



compete in markets located in other countries of the world

For the buyers, import of goods produced in another country is one way of expanding the choice of goods beyond what is domestically produced.

world

-

-



**PRINT CULTURE AND THE
MODERN WORLD**

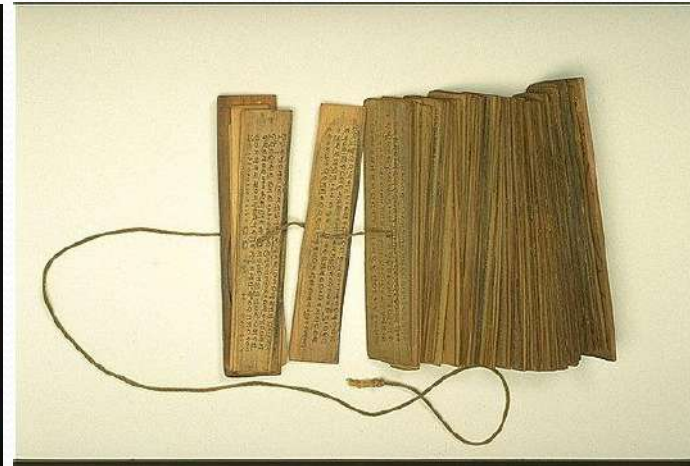
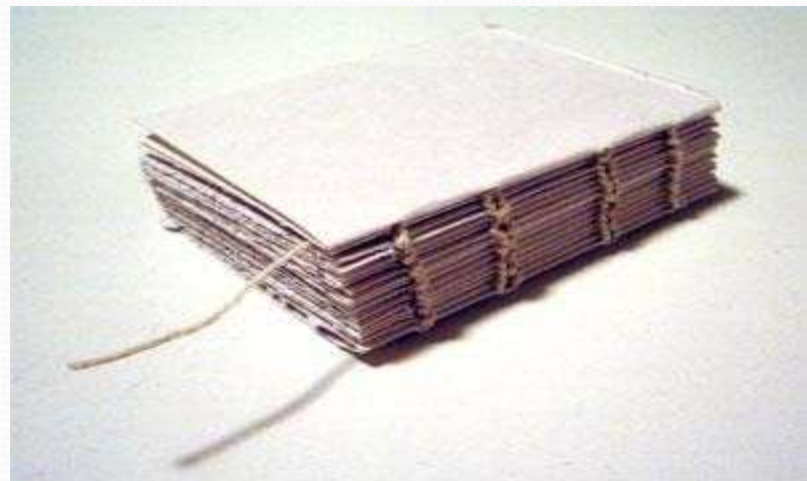


THE FIRST PRINTED BOOKS

- ❖ **The Earliest kind of print technology developed in China, Japan and Korea**
- ❖ **From AD 576 onwards, books in China were printed by rubbing paper against inked surface of woodblocks**

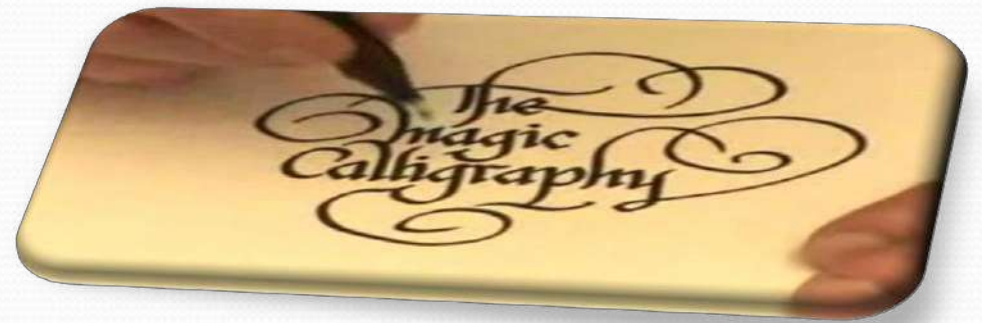
ACCORDION BOOK

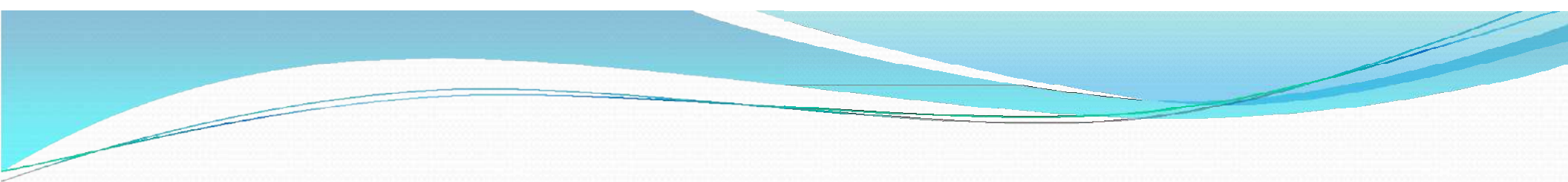
❖ The Traditional Chinese ‘accordion book’ was folded and stitched at the side



CALLIGRAPHY

- ❖ **Calligraphy is the art of beautiful and stylised writing**
- ❖ **Superbly skilled craftsmen could duplicate it with accuracy**





❖ For a very long time china remained the major producer of printed material

❖ Further the Civil service examination expanded the use of print material

❖ Apart from scholars even merchants started using print material



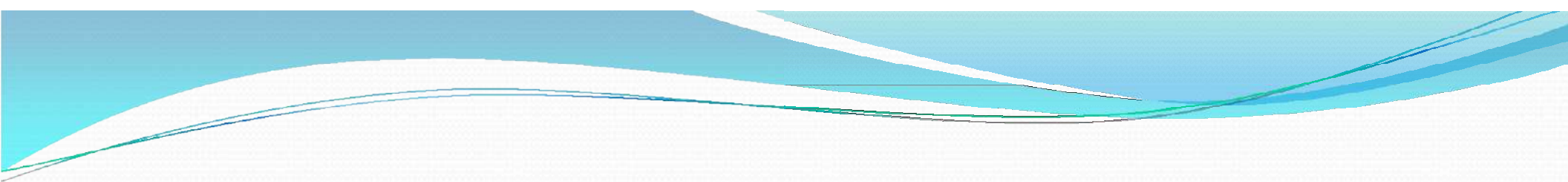
- ❖ **Rich women began to read and publish their poetry and plays**
- ❖ **New reading culture was occupied by new technology**
- ❖ **Shanghai became the hub of the new print culture**

PRINTING IN JAPAN



❖ Buddhist missionaries from China introduced hand-printing technology into Japan (AD 768- 770)

❖ The oldest Japanese book, printed in AD 868, is the Buddhist 'DIAMOND SUTRA'

- 
- ❖ **Pictures were printed on textiles, playing cards and paper money**
 - ❖ **In medieval Japan, poets and prose writer were regularly published**
 - ❖ **Books were cheap abundant**
 - ❖ **Printing of visual material led to interesting publishing practices**

PRINT COMES TO EUROPE

- ❖ **In 1295, Marco polo, a great explorer returned to Italy after many years of exploration in China**
- ❖ **He brought the knowledge of print technology back with him from China**
- ❖ **Luxury editions were still hand written on very expensive VELLUM**

LIMITATIONS



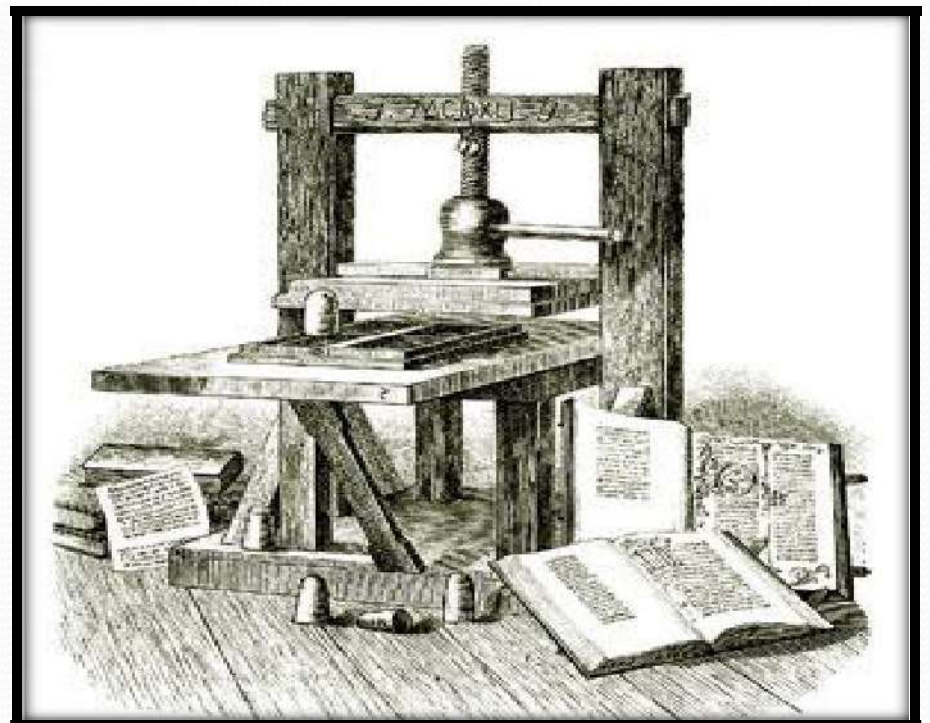
- ❖ **Handwritten manuscripts could not satisfy the ever- increasing demands for books**
- ❖ **Copying was expensive, laborious and time-consuming**
- ❖ **Manuscripts were fragile, difficult to handle and carry around**
- ❖ **Their circulation remained limited**
- ❖ **Thus there was a great need for quicker and cheaper production**



RISE OF PRINTING PRESS

- ❖ Gutenberg was the son of a merchant and grew up on a large agricultural estate**
- ❖ He became a master goldsmith**
- ❖ He created lead moulds for making trinkets**
- ❖ He adopted this technology to design new innovation**

GUTENBERG'S PRINTING PRESS



THE BIBLE





PRINT REVOLUTION (MEANING)

- ❖ **Development of new ways of producing books**
- ❖ **Transformed the lives of people**
- ❖ **Change in their relationship with institutions and authorities**
- ❖ **Influenced popular perceptions**
- ❖ **Opened up new ways of looking at things**

IMPACT OF PRINT REVOLUTION

A.

- **ANEW READING PUBLIC**

B.

- **RELIGIOUS DEBATES AND THE FEAR OF PRINT**

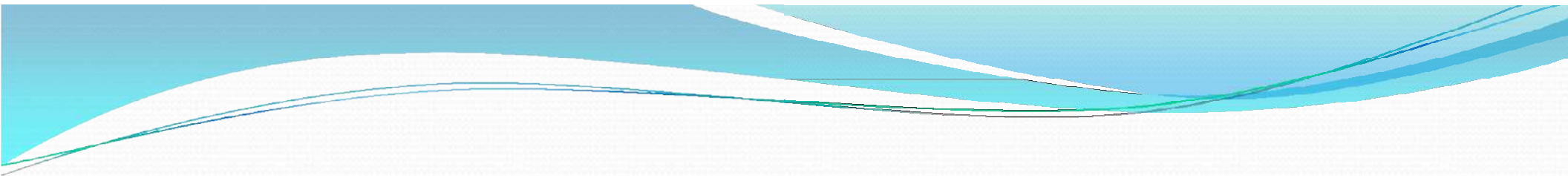
C.

- **PRINT AND DISSENT**



AS A RESULT....

- ❖ **Oral culture entered print and printed material was orally transmitted**
- ❖ **The hearing public and reading public became intermingled**



B. RELIGIOUS DEBATES AND THE FEAR OF PRINT

- ❖ **Print created the wide circulation of ideas**
- ❖ **Introduced a new world of debate and discussion**



FEAR OF PRINT...

- ❖ Many were apprehensive of the effect of wider circulation of books on the mind of people
- ❖ Rebellious & irreligious thoughts might spread
- ❖ The authority of valuable literature would be destroyed
- ❖ this anxiety to the widespread criticism of print media



EXAMPLE...

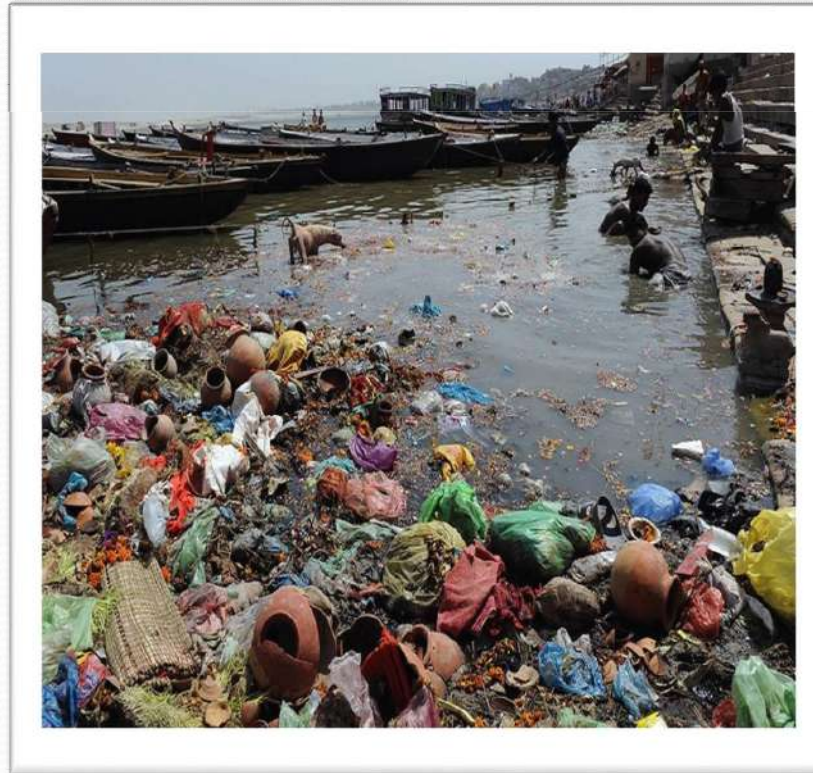
- ❖ **In 1517, the religious reformer Martin Luther wrote *Ninety five theses* criticising many of the practices & rituals of the Roman Catholic Church**
- ❖ **This led to a division within the church and to the beginning of the ‘Protestant Reformation’**



C. PRINT AND DISSENT

- ❖ **Print & religious literature stimulated many distinctive individual interpretations of faith**
- ❖ **Manocchio reinterpreted the message of Bible and formulated a view of god & creation that enraged the Roman Catholic Church**

Today, it is considered to be the fifth most polluted river in the world.





Reasons for the pollution of Ganga

Human
Waste

Industrial
Effluents

Religious
Ceremonies

AGAINST WATER POLLUTION IN INDIA

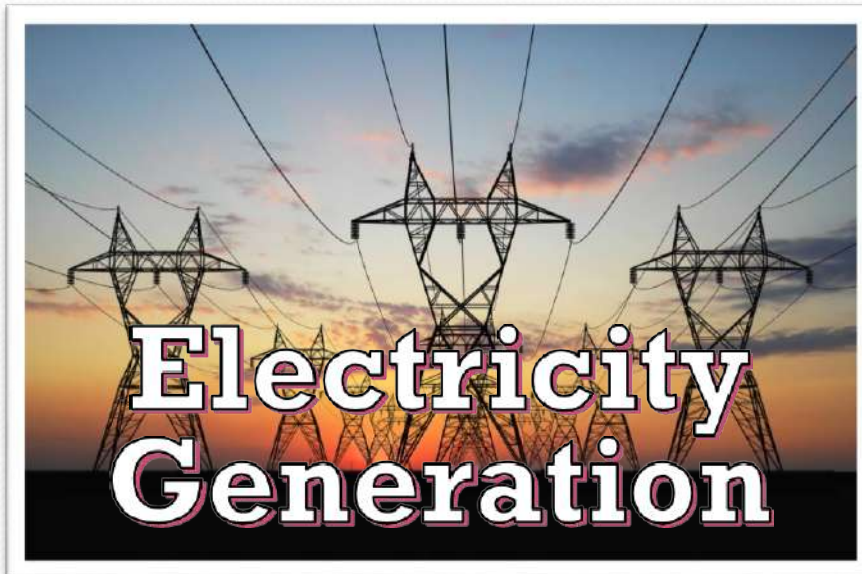
Narmada Bachao Andolan

Save Ganga Movement

Tehri Dam Andolan



USES OF DAMS



CONSEQUENCES WHEN A DAM IS BUILT...

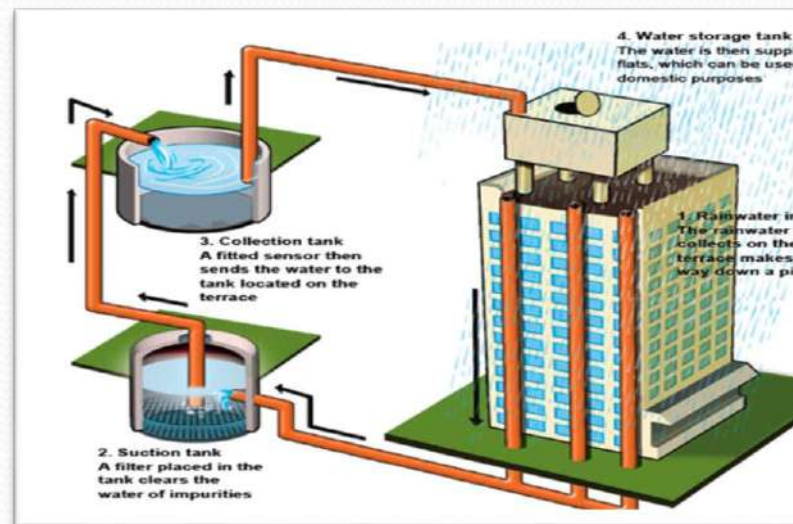
Regulating and damming of rivers affect their natural flow causing poor sediment flow and excessive sedimentation at the bottom of the reservoir resulting in rockier stream beds and poor habitat for aquatic life's

Make it difficult for aquatic fauna to migrate, especially for spawning.

Triggers flood

RAIN WATER HARVESTING

Rainwater harvesting is a technology used for collecting and storing rainwater from rooftops, the land surface or rock formations using simple techniques such as jars and pots as well as more complex techniques such as underground check dams.





PHYSICS

What is Light?

Light is a wave, or rather acts like a wave.

How do we know?

- Reflection
- Refraction
- Dispersion
- Diffraction
- Interference
- Polarization



Electromagnetic Waves

- Electromagnetic waves are special in the fact that they do not need a medium to propagate through.
- But what is creating the disturbance? What is emitting this energy?
- **ELECTRONS**



Electromagnetic Waves

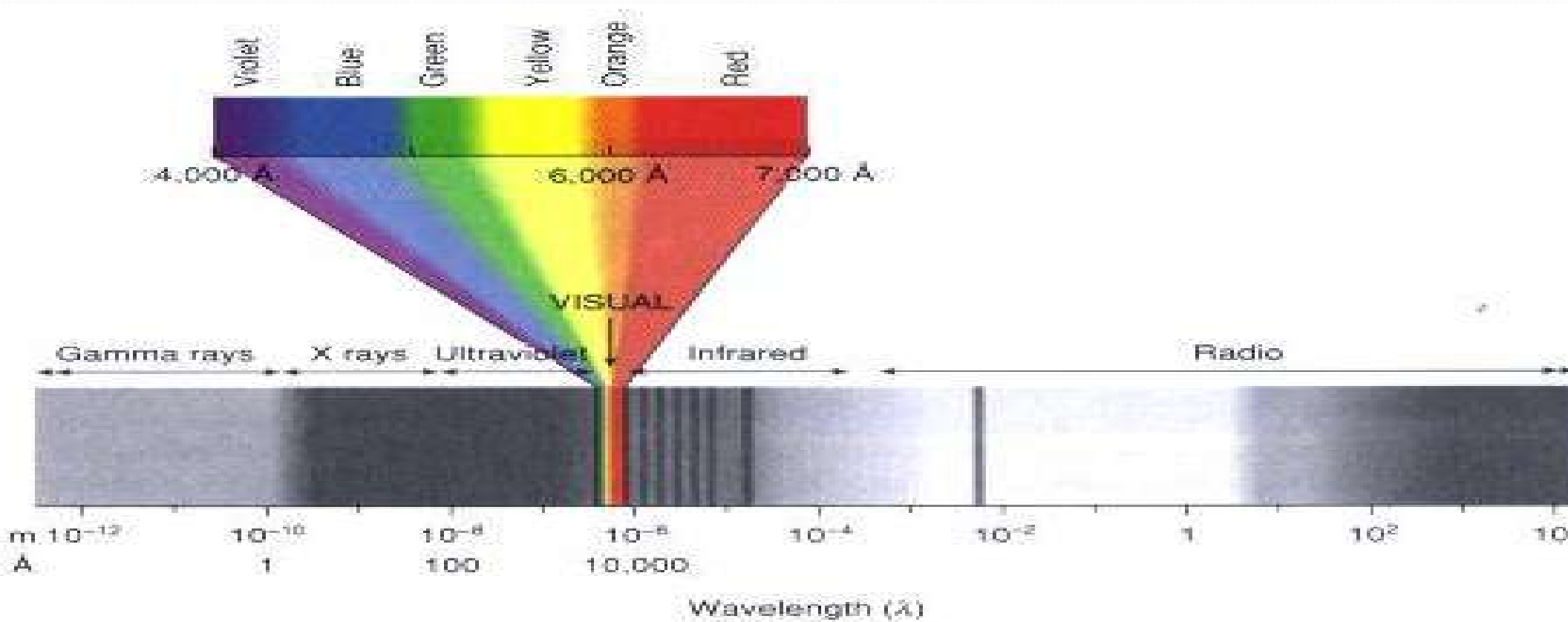
- Electrons in materials are vibrated and emit energy in the form of **photons**, which propagate across the universe.
- Photons have no mass, but are pure energy.
- Electromagnetic Waves are waves that are made up of these “**photons**”.
- When these photons come in contact with boundaries, E-M waves interact like other waves would.



Electromagnetic Waves

- Electromagnetic waves are everywhere.
 - Light is only a small part of them
-
- **Radios**
 - **TVs**
 - **Microwaves**
 - **Light**
 - **Radiation**
 - **Lasers**
 - **CD/DVD players**
 - **X-Rays**

Electromagnetic Spectrum





Speed of E/M Waves

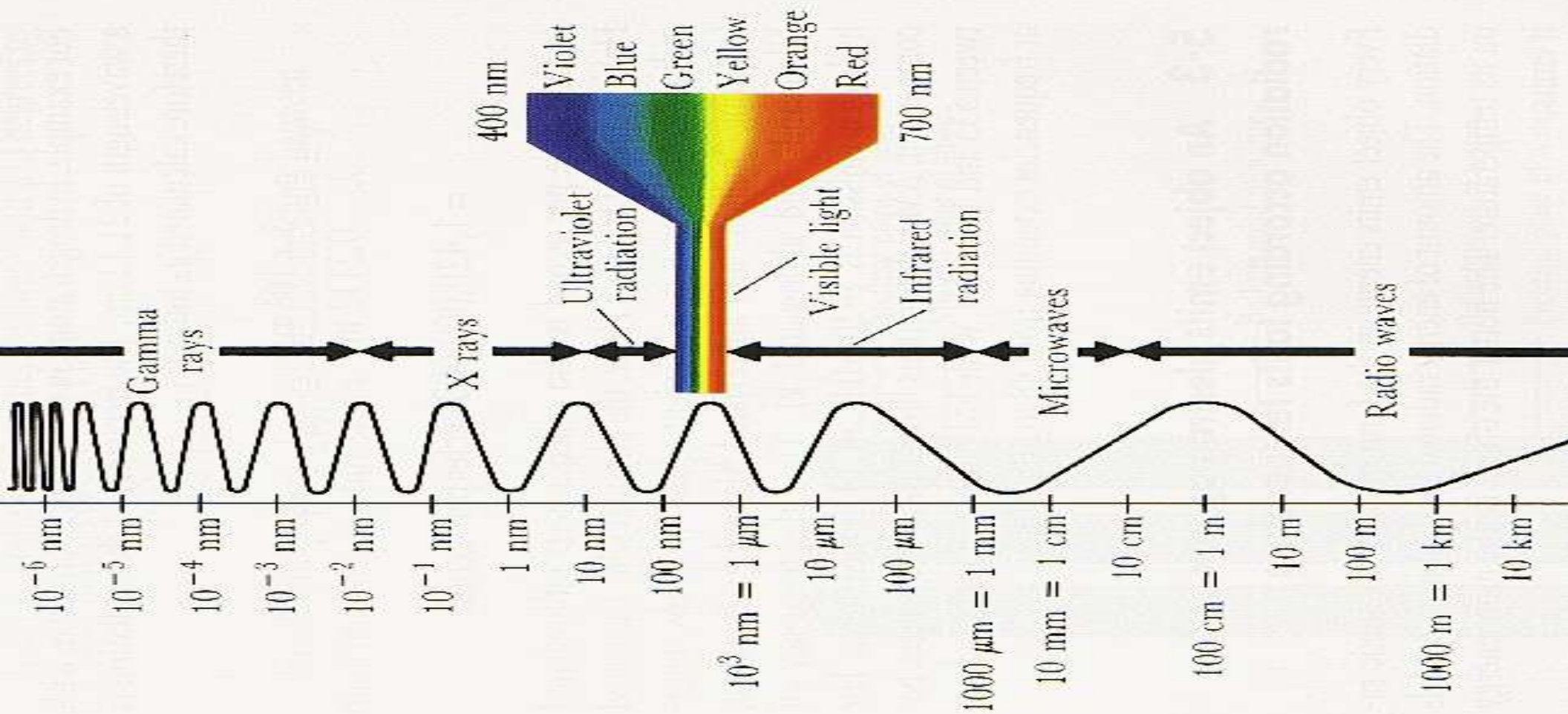
- From last chapter, we found that
 - $V = f * \lambda$
- We also said that the speed of a wave in a certain medium is always constant.
- It has been found that the speed of E-M waves and light is ---
 - 3×10^8 or 300,000,000 m/s
 - 671,000,000 mph
 - 186,000 miles per second
 - We call this value “c”


$$c = f * \lambda$$

- c is **constant** throughout the universe, as long as light is in vacuum.
- When it is in other materials, c can change, but can never be larger than its value in a vacuum.
- Since “ c ” is constant, all of E-M waves will have a corresponding **frequency** to go along with their **wavelength**.

$$c = f * \lambda \quad \sim \sim \sim \quad f = c / \lambda$$

Lets find the corresponding frequency ranges for a few of the groups of E-M waves.





Energy in E-M Waves

- Which waves have more energy, Radio waves or gamma waves?
- The **greater the frequency** of an E-M wave, the more crests pass a point in a certain amount of time, therefore the more photons pass that point.
- This means that **more energy** moves past that point in a certain amount of time or that the wave contains **more energy**.

The Human Eye

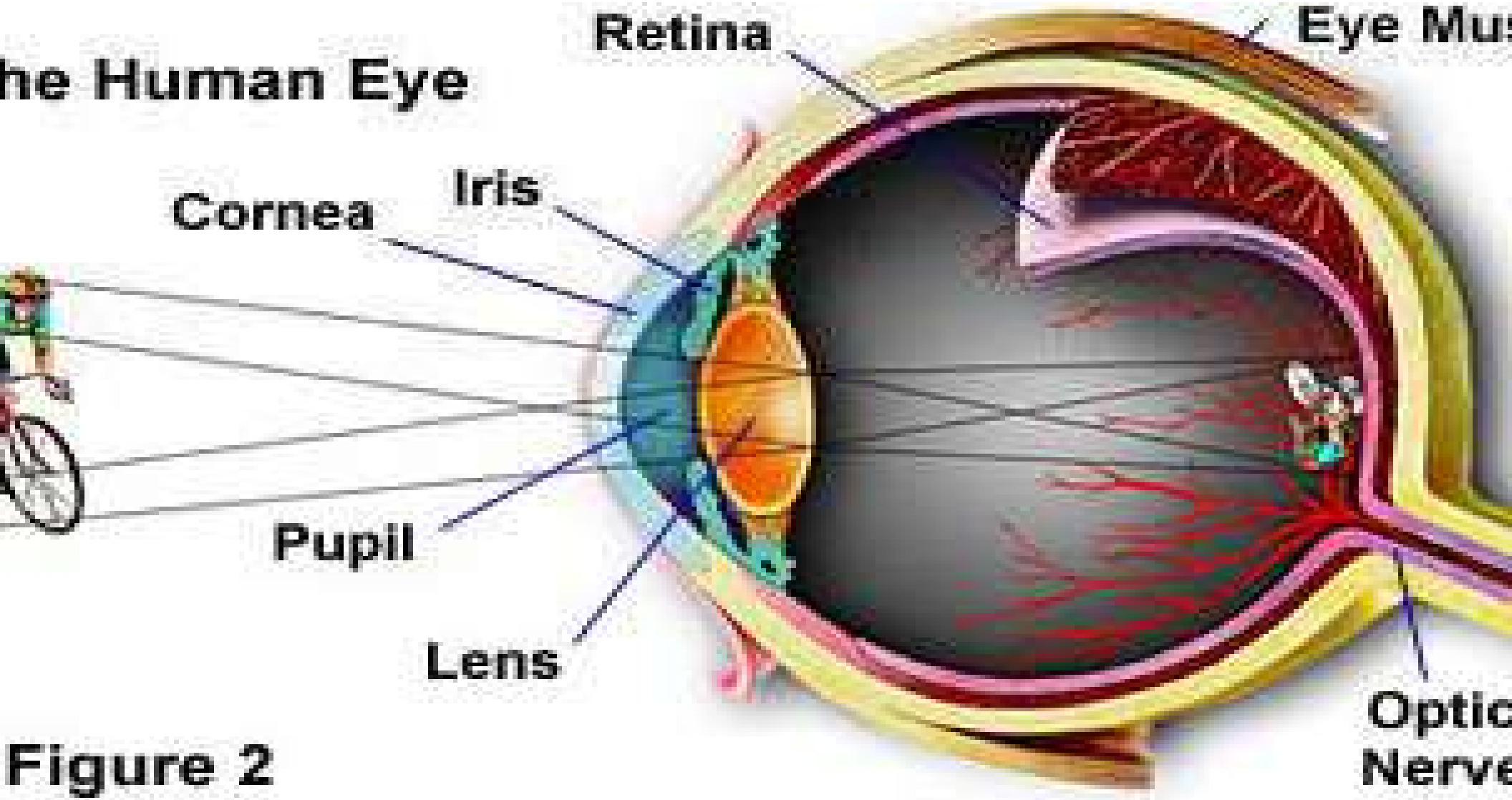


Figure 2

Optic Nerve



Visible Light

We now know what we see is part of the electromagnetic spectrum. We know that the light waves enter our eye, and stimulate parts of it that cause a electrical impulse to be sent to the brain which creates this visual image.

But everything does not emit radiation. How do we see those things?
And why cant we see a window?



Seeing things

We know that when waves run into a boundary they are partially transmitted and partially reflected.

Light behaves as a wave, so it too is reflected.

Therefore, an object does not need to emit photons itself to be seen. It just has to reflect light back to our eyes where we can detect it.

Objects that do not allow light to pass through them are called **opaque**.

Objects that allow light to pass through them are considered **transparent**.

Objects in between are called **translucent**.

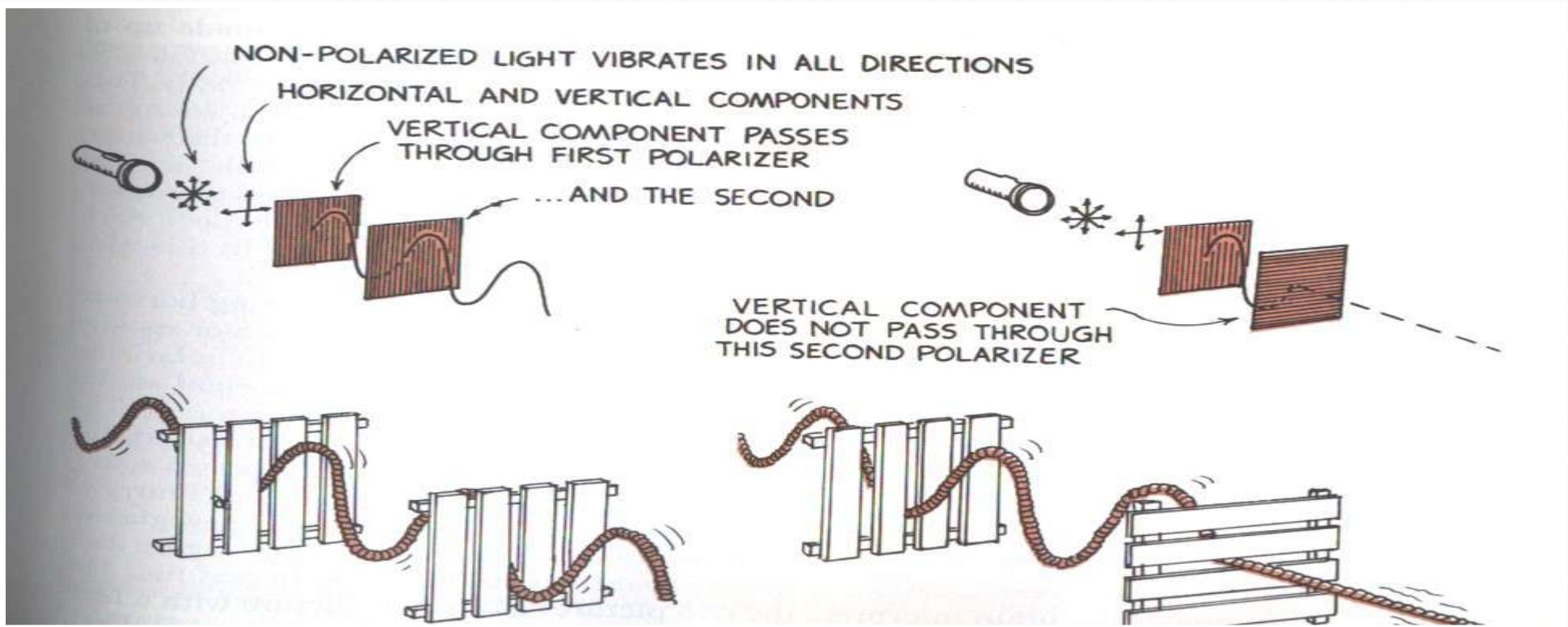


Polarization

Polarization is a phenomenon of light that is used in sun-glasses and 3-D movies.

Play with the two polarizing filters for a few minutes and note what is happening and see if you can think of any reasons for it.

Polarization





Color

Different objects may emit different wavelengths of E-M radiation, so we would see that light as different colors.

But why do we see colors in objects that reflect light? If you shine a white light on my clothes, and it gets reflected why doesn't all of my clothes appear white?

When I shine white light through a colored piece of plastic, why does it change color?



Color

The light we see is known as visible or white light – although it is not that simple.

The light is not really white, the white we see is a combination of all the colors of the rainbow.

Remember **R-O-Y G. B-I-V** from art class.

When all of these light waves are combined we see white light.



Color Reflection

So if we see something as **WHITE**, that means ...

- It reflected back all the wavelengths of light to our eyes

If we see something as **RED** or **BLUE**

- It reflected only the **RED** or only the **BLUE** wavelengths
- The others were absorbed.

And if we see something as black?

- It did not reflect back any of the light.



Color Transmission

Filters work in a similar way.

- **Red** filters only let RED light thru.
- **Blue** let only BLUE light thru.

What do you think that UV sticker means on your sunglasses?

Why do they sell those orange glasses that are supposed to reduce glare?



Some Sweet Color Tricks

Combining colors in art class

How does [color printing work](#)?

Combining lights

Why is the sky blue?

Why are sunsets red?

Why is water greenish-blue?

[How does 3-D work](#)?

Why does a CD reflect a rainbow, and a mirror does not?

How can you help people who are color blind?

OTHERS ← [link to site](#)



UX

- We now know how light behaves, but we must measure how strong it is.
- The rate at which a source emits light is called the **LUMINOUS FLUX (P)**.
- What do you think this is measured in? What are light bulbs measured in.
- **LUMINOUS FLUX (P)** is actually measured in something called a **lumen (lm)**.
- A typical 100-W bulb emits 1750 lm.



uminance

Flux is the total of all the light that is emitted from a source.

This is not very useful, often we would like to know how much of that light is hitting a surface at some point.

The illumination of a surface is called **illuminance, E**. It is measured in lumens per square meter, lm/m^2



illuminance

How do you think illuminance is affected when the object moves away from the source?

- Right the illuminance decreases

So what would you expect an equation to look like for E in terms of P and the distance away d ?

- Close it is actually

$$E = \frac{P}{4\pi d^2}$$

CHEMISTRY
CHAPTER - 5

PERIODIC CLASSIFICATION
OF ELEMENTS

1) Classification of elements :-

The arranging of elements into different groups on the basis of the similarities in their properties is called classification of elements.

The classification of similar elements into groups makes the study of elements easier.

There are about 114 different elements known so far.

Early attempts at classification of elements :-

The earliest attempt to classify elements was grouping the then known elements (about 30 elements) into two groups called metals and non metals.

The defect in this classification was that it had no place for metalloids (elements which have properties of both metals and non metals) which were discovered later.

b) Dobereiner's Triads :-

Dobereiner classified elements in the increasing order of their atomic masses into groups of three elements called triads. In each triad the atomic mass of the middle element was approximately equal to the average atomic mass of the other two elements.

A defect in this classification was that all the then known elements could not be correctly arranged into triads.

Triad		Atomic mass	Average atomic mass of 1 st and 3 rd element
Lithium	Li	6.9	22.95
Sodium	Na	23.0	
Potassium	K	39.0	
Calcium	Ca	40.1	88.7
Strontium	Sr	87.6	
Barium	Ba	137.3	
Chlorine	Cl	35.5	81.2
Bromine	Br	79.9	
Iodine	I	126.9	

c) Newland's octaves :-

Newland classified the elements in the increasing order of their atomic masses into groups of eight elements called octaves like the notes of music. He found that when the elements are arranged in the increasing order of their atomic masses into octaves then there was a similarity of properties in every eighth element.

	re	ga	ma	pa	da	ni
	Li	Be	B	C	N	O
	Na	Mg	Al	Si	P	S
	K	Ca	Cr	Tl	Mn	Fe
and Ni	Cu	Zn	Y	In	As	Se
r	Rb	Sr	Ce and La	Zr	-	-

defect in this classification was:-

the known elements and elements discovered later could not be correctly arranged in octaves.

Some elements having different properties were placed in the same rows like cobalt and nickel having different properties are placed along with Fluorine, Chlorine and Bromine. Elements having properties similar to Cobalt and Nickel are placed in different rows.

3a) Mendeleev's periodic law :-

Mendeleev's periodic law states that, 'The properties of elements are periodic functions of their atomic masses'.

B	A B	A B	A B	A B	A B	A B	
							Transition series

	II	III	IV	V	VI	VII	VIII		
11 Li 7.0	4 Be 9.01	5 B 10.8	6 C 12.0	7 N 14.0	8 O 16.0	9 F 19.0			
12 Mg 24.3	13 Al 27.0		14 Si 28.1	15 P 31.0	16 S 32.1	17 Cl 35.5			
19 K 39.1	20 Ca 40.1		22 Ti 47.9	23 V 50.9	24 Cr 52.0	25 Mn 54.9	26 Fe 55.9	27 Co 58.9	28 Ni 58.7
29 Cu 63.5	30 Zn 65.4			33 As 74.9	34 Se 79.0	35 Br 79.9			
37 Rb 85.5	38 Sr 87.6	39 Y 88.9	40 Zr 91.2	41 Nb 92.9	42 Mo 95.9		44 Ru 101	45 Rh 103	46 Pd 106.4
47 Ag 108	48 Cd 112	49 In 115	50 Sn 119	51 Sb 122	52 Te 128	53 I 127			
55 Cs 133	56 Ba 137	57 La 139		73 Ta 181	74 W 184		76 Os 194	77 Ir 192	78 Pt 195.1
79 Au 197	80 Hg 201	81 Tl 204	82 Pb 207	83 Bi 209					
			90 Th 232		92 U 238				

4a) Modern periodic law :-

Modern periodic law states that, 'The properties of elements are periodic functions of their atomic numbers'.

2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17
Group →

Periodic Table of the Elements

										IIIA	IVA	VA	VIA	VIIA	
4 Be											5 B	6 C	7 N	8 O	9 F
12 Mg	III B	IV B	V B	VIB	VII B	VIII B			IB	IIB	13 Al	14 Si	15 P	16 S	17 Cl
20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br
38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I
56 Ba	57 *La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At
88 Ra	89 +Ac	104 Rf	105 Ha	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Uuu	112 Uub	113 Uut				
Lanthanide Series	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu	
Actinide Series	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr	

Periodic Table of the Elements

- hydrogen
- alkali metals
- alkali earth metals
- transition metals
- poor metals
- nonmetals
- noble gases
- rare earth metals

4											5	6	7	8	9
Be											B	C	N	O	F
12											13	14	15	16	17
Mg											Al	Si	P	S	Cl
20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35
Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br
38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53
Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I
56	57	72	73	74	75	76	77	78	79	80	81	82	83	84	85
Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At
88	89	104	105	106	107	108	109	110							
Ra	Ac	Unq	Unp	Unh	Uns	Uno	Une	Unn							

58	59	60	61	62	63	64	65	66	67	68	69	70	71
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
90	91	92	93	94	95	96	97	98	99	100	101	102	103
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

5. Properties of elements in periods and groups :-

Valence electrons :-

In a period the number of valence electrons increases from 1 to 8 from the left to the right. The number of shells is the same.

2nd Period

Elements	- Li,	Be,	B,	C,	N,	O,	F,	Ne
	- 3	4	5	6	7	8	9	10
	- 2,1	2,2	2,3	2,4	2,5	2,6	2,7	2,8
Valence electrons	- 1	2	3	4	5	6	7	8
	- 2	2	2	2	2	2	2	2

In a group the number of valence electrons is the same for all the elements but the number of shells increases from top to bottom.

Group - I A

Elements	AN	EC	VE	Shells
	1	1	1	1
	3	2,1	1	2
	11	2,8,1	1	3
	19	2,8,8,1	1	4

ii) Valency :-

period the valency of the elements increases from 1 to 4 and then decreases from the left to the right.

2nd Period

Elements	Li	Be	B	C	N	O	F	Ne
Atomic Number	3	4	5	6	7	8	9	10
Electron Configuration	2,1	2,2	2,3	2,4	2,5	2,6	2,7	2,8
Valence Electrons	1	2	3	4	5	6	7	8
Valency	1	2	3	4	3	2	1	0

In a group the valency is the same for all elements of the group.

Group - I A

Elements	AN	EC	VE	Valency
H	1	1	1	1
Li	3	2,1	1	1
Na	11	2,8,1	1	1
K	19	2,8,8,1	1	1

iii) Atomic size (Radius of the atom) :-

In a period the atomic size of the elements decreases from the left to the right because the nuclear charge (number of protons) increases and so the electrons are pulled closer to the nucleus.

2nd Period

Elements	- Li,	Be,	B,	C,	N,	O,	F,	Ne
	- 3	4	5	6	7	8	9	10
	- 2,1	2,2	2,3	2,4	2,5	2,6	2,7	2,8
protons	- 3	4	5	6	7	8	9	10



 Atomic size decreases

In a group the atomic size of the elements increases from top to bottom because the number of shells increases and the distance between the nucleus and shells also increases.

Group - I A

Elements	AN	EC	VE	Shells	Atomic size
H	1	1	1	1	increases
Li	3	2,1	1	2	
Na	11	2,8,1	1	3	
K	19	2,8,8,1	1	4	

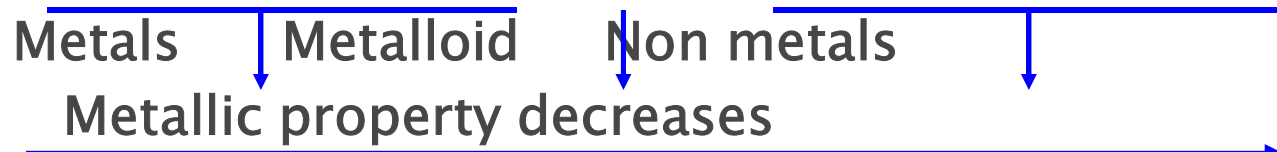


iv) Metallic property (Electropositive nature) :-

In a period the metallic property of the elements decreases from the left to the right.

3rd Period

Elements - Na, Mg, Al, Si, P, S, Cl, Ar



In a group the metallic property of the elements increases from the top to the bottom.

Group VI A

Elements

Carbon	C	-	Non metal	Metallic property increases
Silicon	Si	-	Metalloid	
Germanium	Ge	-	Metalloid	
Tin	Sn	-	Metal	
Lead	Pb	-	Metal	

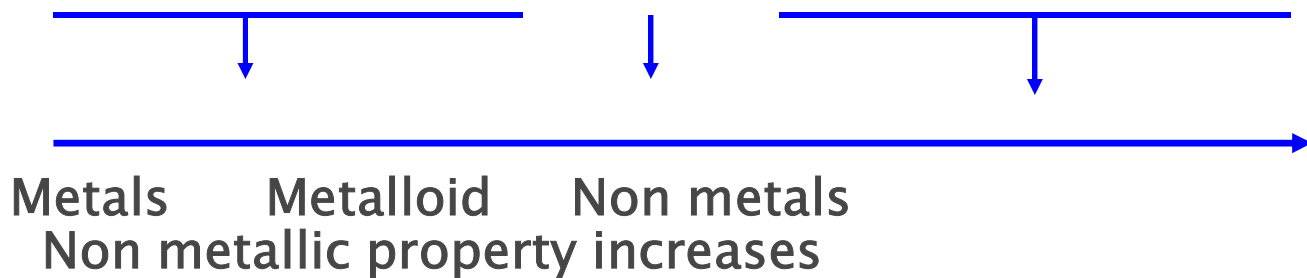


v) Non metallic property (Electronegative nature) :-

period the non metallic property of the elements increases from left to the right.

3rd Period

Elements - Na, Mg, Al, Si, P, S, Cl, Ar



group the non metallic property of the elements decreases from top to the bottom.

- Group VI A

Elements				
Carbon	C	-	Non metal	Non metallic property decreases
Silicon	Si	-	Metalloid	
Germanium	Ge	-	Metalloid	
Tin	Sn	-	Metal	
Lead	Pb	-	Metal	





THANK YOU